

oxime, m.p. (from methanol) 184-185°, $[\alpha]_D -17^\circ$ (c , 0.71) (15.2 g.). The structure of the oxime was proved by a series of transformations analogous to those described above for the first example of the reaction.

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A SYNTHESIS OF ALDOSTERONE ACETATE

Sir:

We wish to report a convenient three step partial synthesis of aldosterone acetate¹ using the new photochemical reaction reported in the preceding communication.² Corticosterone acetate (I, R = H, X = H₂) in dry pyridine was treated with excess nitrosyl chloride at room temperature to give corticosterone acetate nitrite (I, R = NO, X = H₂), m.p. 174-176°, $[\alpha]_D +316^\circ$ (c , 1.1; all rotations in CHCl₃). This nitrite (4.0 g.) in toluene (200 ml.) was irradiated at 32° under pure nitrogen for 75 min. as described.² The crystalline solid which had separated was removed (885 mg., 21.2%) and identified as aldosterone acetate oxime (I, R = H, X = NOH). Recrystallized from benzene this had m.p. 175-194°, $[\alpha]_D +198^\circ$ (c , 1.3), λ_{\max} 240 m μ ($\epsilon = 16,500$), $\nu_{\max}^{\text{CHCl}_3}$ 3550, 3350, 1740, 1665 and 1615 cm.⁻¹. Treatment with pyridine-acetic anhydride at 100° for 5 min. gave the oxime acetate (I, R = N, X = NOAc), m.p. (from ethyl acetate) 183-187°, $\nu_{\max}^{\text{CHCl}_3}$ 3650, 1780, 1750, 1675 and 1625 cm.⁻¹. Aldosterone acetate oxime (505 mg.) was added at 10° to a mixture of acetic acid (8 ml.) and aqueous sodium nitrite (5%, 4 ml.) and kept with agitation for 5 min. Extraction with methylene chloride gave, on crystallization from ethyl acetate, aldosterone 21-acetate (II) (320 mg.), identified by m.p., rotation, analysis, ultraviolet and infrared spectra (comparison with authentic racemate) and by paper chromatography.

On melting, or on refluxing in methanol for 1 hr., aldosterone acetate oxime was converted into

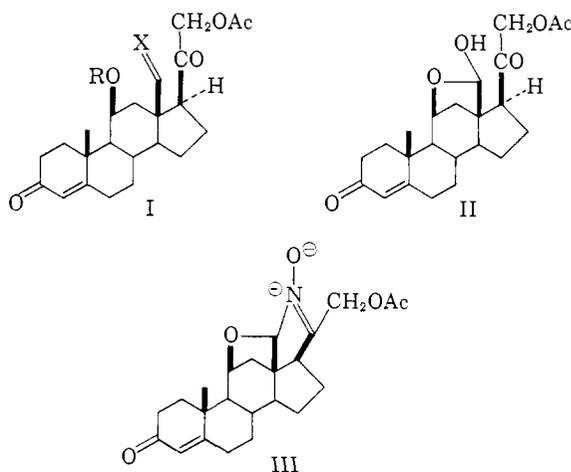
(1) S. A. Simpson, T. F. Tait, A. Wettstein, R. Neher, J. V. Euw, O. Schindler and T. Reichstein, *Helv. Chim. Acta*, **37**, 1163, 1200 (1954).

(2) D. H. R. Barton, J. M. Beaton, L. E. Geller and M. M. Pechet, *This Journal*, **82**, 2640 (1960).

the nitron (III), m.p. 194-197°, $[\alpha]_D +119^\circ$ (c , 1.0), λ_{\max} 239 m μ ($\epsilon = 27,400$), ν_{\max}^{KBr} 1735, 1660 and 1600 cm.⁻¹. The usual infrared band for a 20-ketone was absent. We shall discuss the mechanism of the nitrite photolyses reported here and in the preceding communication² in our complete paper.

Very recently an alternative partial synthesis of aldosterone has been reported involving about 20 steps from 3 α -acetoxypregnane-11,20-dione.³

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(3) K. Heusler, J. Kalvoda, C. Meystre, P. Wieland, G. Anner, A. Wettstein, G. Cainelli, D. Arigoni and O. Jeger, *Experientia*, **16**, 21 (1960). See also L. Velluz, G. Muller, R. Bardoneschi and A. Poitvein, *Compt. rend.*, 725 (1960).

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STRUCTURE OF O,O'-DIETHYL METHYLPHOSPHONOTHIOATE AND CONJUGATIVE PROPERTIES OF THE P=S BOND

Sir:

In connection with more comprehensive studies on the nature of heteroorganic bonds, we have examined recently the structure of O,O'-diethyl methylphosphonothioate, CH₃P(S)(OCH₂CH₃)₂, and have ascertained a novel structural feature which has important implications for reaction mechanisms involving this and analogous organic phosphorus systems. Employing high-resolution nuclear resonance equipment,¹ we have obtained fine details of both the H¹ and P³¹ spin-resonance transitions of this compound in an extremely homogeneous magnetic field of 14,092 gauss. Two fixed-frequency oscillators were used successively to irradiate the sample. The material was especially purified for this and other studies.

In the H¹ trace, the two expected higher-field band patterns stand out clearly: an equal inten-

(1) Varian Associates V4300-2.